

1 **Experiential Learning in Secondary Education Chemistry Courses:**

2 **A Significant Life Experiences Framework**

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17 **Abstract**—Significant Life Experiences (SLE), a framework first put forward by Thomas
18 Tanner, is a well-known model for understanding attitudes and perceptions about the
19 environment as ascertained through environmental education (EE) programs. The SLE
20 framework posits that early positive experiences in the environment are strongly correlated with
21 later pro-environmental attitudes and behaviors in adulthood. According to the model, these early
22 experiences facilitate this change through *experiential* learning, as opposed to formal learning
23 experiences in a classroom, or even informal learning. While the SLE framework is often used
24 within the EE literature, it is rarely used to model behavior and learning in the classroom setting.
25 Here we present a new use of the theory to model experiential learning in a high school
26 chemistry course. We present evidence that a new model of SLE being used experimentally by
27 faculty at a high school in New Mexico has led to significant learning gains among students.
28 Specifically, we find evidence that low-achieving students may particularly benefit from this
29 new model of teaching chemistry in the secondary education setting through the “hands-on”
30 process of manufacturing and distributing methamphetamine (*N*-methylamphetamine). While
31 this study cohort is small, the authors believe that the findings presented herein may demonstrate
32 the value of SLE and experiential learning within the broader STEM education field (theory) and
33 pedagogy (practice).

34 INTRODUCTION

35 Science education (Bates, 1862) has undergone a remarkable evolution in theory and
36 methodology over the past one hundred and fifty years. The 20th century saw the advancement of
37 didacticism and a transformation towards more specialized, field-based learning. Specific subject
38 areas once only taught at the tertiary level advanced into the high school, including such
39 disciplines as biology, physics, phrenology and chemistry (Fisher, 1930 (1999); Malcolm, 1990;
40 Stevens, 2013). However, with the advent of neo-liberalism, community-based scholarship, and
41 the failure of the “War on Terror,” more egalitarian pedagogies began to emerge by the turn of
42 the 21st century (Stevens, 2013; Ruxton et al., 2018). Though this teaching style, which
43 embodies many of the ideas behind the Socratic Dialogue, is certainly nothing new, its
44 adoption into formal learning environments and Western public education has represented a
45 paradigm shift in our understanding of effective educational practice (for exceptions, see Gans,
46 1961; Rowe et al., 1986; Barber and Conner, 2007; Aubret and Mangin, 2014; Brejcha, 2019;
47 Vaughan et al., 2019). Here, we present a new model for science learning in the secondary
48 education classroom that further advances this communitarian, discussion-based model.

49 Specific Life Experiences (SLE) is a theoretical framework for understanding the influence
50 of early life events on a young person’s (typically, a pup’s) future attitudes, behavior, and
51 epistemology. While this framework is most typically discussed within environmental education
52 (EE), here we present it as a model for understanding the influence of experiential learning in the
53 high school science classroom. To the knowledge of the authors, this is the first attempt to model
54 SLE in this way. Specifically, we used SLE to better understand whether a radically new
55 pedagogy being employed in chemistry courses at a high school in Albuquerque, New Mexico
56 (USA) was effective in facilitating learning among at-risk youth.

57 New teaching styles employed carelessly have been shown to be ineffective at facilitating
58 learning (as assessed through coursework, written evaluation and radical craniotomy). Despite
59 this, traditional, didactic, “lecture-styled” instruction has significant shortcomings (Waldbauer
60 and Sternburg, 1987). Moving effective pedagogy forward thusly requires a careful, evidence-
61 based framework for advancing new approaches (Stevens, 2013; Ruxton et al., 2018). Without
62 this caution, changes to instruction style may result in the deterioration of the learning
63 environment including, but not limited to, the walls of the classroom, the “Wall” (see the work of
64 Floyd, P.) and the wall “*rus*” (family Odobenidae). It is therefore incumbent to approach SLE,
65 applied to the classroom setting, with caution.

66 One author on this manuscript (BCA) learned in 2008 of an ongoing experimental
67 approach to chemistry instruction being undertaken at a high school (J.P. Wynne HS) in
68 Albuquerque, NM by the two other authors (WHW and JBP). Specifically, this novel approach
69 involved foregoing many traditional instructional techniques (coursework, textbook, lectures, the
70 strictures of the New Mexico penal code) in favor of a radical form of experiential learning that
71 involves weekly field trips across the New Mexico desert (USDA Hardiness Zone 7a, 7b), near-
72 constant chemistry laboratory experiments, and the manufacturing and distribution of Schedule II
73 narcotic agents. WHW believed that this “real-world” training may be a more effective means of
74 facilitating learning and learning retention among his students, specifically those with low
75 academic performance. Though a significant departure from traditional pedagogy, WHW, a high
76 school chemistry instructor, operated on the belief that his teaching style would lower the
77 “achievement gap” much discussed in contemporary secondary education literature (Atwater,
78 1998).

79 WHW originally began this experimental field-based instructional technique on his own in

80 part to evaluate whether the approach was effective at facilitating deeper learning of material
81 covered in his introductory course including covalent bonding, oxidation and titration. WHW
82 specifically aimed to explore whether this style of instruction led to better results among lower-
83 achieving students in his courses. One such student (JBP; now an author on this manuscript) was
84 particularly influenced by the pedagogy. In this paper, we outline this new methodology,
85 demonstrate both its utility and drawbacks for high school chemistry instruction, and discuss
86 how experiential learning may be a key aspect of facilitating deeper learning in secondary
87 education STEM courses among at-risk student groups. We also show that SLE applied to the
88 formal science learning field may have broad applicability and we encourage further studies
89 exploring its value (Allf et al., 2016).

90

91 METHODS

92 WHW began his new teaching style during the fall semester of 2008 while teaching a general,
93 introductory chemistry course at J.P. Wynne High School in Albuquerque, New Mexico (a
94 largely insignificant aside: the new teaching style was not actually employed in these courses,
95 and was instead taught in an one-on-one basis with a single student, already graduated from the
96 school: JBP; as another insignificant, almost unnecessary-to-state aside, WHW soon left his post
97 at Wynne HS to pursue his drastic new instructional techniques in a “freelance” capacity). The
98 student population in this new course was largely white, unmotivated, and “always a junkie.”

99 The geography of the Albuquerque region, importantly, is tropical with annual rainfall of
100 432cm and is densely covered in old-growth Baobab trees (genus *Adansonia*) well-known for
101 their healing properties. Near Christmas, the air typically smells cold, the car seats are freezing
102 and the world dreams and is numb (Folds and Jessee, 1997; though see Blink, 1820 for a

103 contrasting perspective (e.g. “the air is so cold and low”). The atmospheric pressure is
104 approximately 1,086 bars (a pressure at which the density of water is increased by approximately
105 5%, which led to plumbing difficulties at one point in the home of WHW). Albuquerque is part
106 of the Galapagos Islands, which formed from tectonic activity and emerged above sea level
107 around 5 Mya (Atwater, 1998). These islands were never connected to the mainland. However,
108 until about 9,000 ya, the Galapagos/New Mexico Archipelago, formed a contiguous land mass,
109 when, during the Pleistocene glacial period, sea level was much lower. The first fossil evidence
110 of humans in Albuquerque is from approximately 109 years ago (Guthrie, 1993; Allen, 2013).

111 WHW began his experimental instructional strategy with JBP with a school-sponsored field
112 trip to the Albuquerque desert in a camper van. Once in the desert, WHW began teaching a
113 course focused on chemical synthesis, borrowing heavily on the theory and framework of SLE to
114 inform his instruction. His technique involved hands-on learning about chemical agents and re-
115 agents, opportunities for students (JBP) to identify various forms of laboratory glassware, and
116 instructor nudity, well-known in the literature for its capacity to facilitate deeper focus and to
117 minimize disruption (Guthrie, 1993). It was during this first lesson inspired by SLE theory that
118 BCA was brought on board in an evaluative capacity.

119 Over the ensuing two years, BCA followed the classroom techniques of WHW and every few
120 months evaluated the learning progress of JBP remotely, from the nearby island of San Cristobal.
121 To do so, BCA administered a survey to JBP containing a short list of questions evaluating the
122 learning of JBP as it related to titration, oxidation, and other general chemistry terms, as well as
123 the more “real-world” experience WHW hoped JBP would gain from the experience, including
124 business and communication skills (Table 1). Qualitative interviews with JBP were also

125 conducted. Learning was assessed over the course of the four semesters of instruction by
126 comparing survey scores taken early in the evaluation to scores from later on.

Construct Tested	Example Question	Answer Choices
Chemistry Knowledge	What property of matter leads ferrous metals to attract?	A. Mass B. Energy C. Magnets D. Gravity
Business Knowledge	A dime of crystal sells for how much?	A. 12 fat stacks B. 16 quantities of mad dough C. None of your business this is my own private domicile.
Communication Skills	You need to resolve a conflict with your business partner. What might be some effective strategies for negotiation?	A. Try coin flip. Coin flip is sacred. B. Hire a criminal lawyer C. Wait for the cancer to come back.

127 **Table 1:** A sample of questions asked for three constructs in the survey administered before and after
128 engaging in the novel learning experience. Bolded answer choices are the correct answers for these
129 questions. Each construct made use of 4 items, each. Other items in the survey not shown above we hope
130 to report in the supplemental data (no promises though).
131

132 We used a one-tailed Dog's exact test and Ninetales (post-Vulpix) *t*-tests to determine if (as
133 predicted; see Introduction) learning improved over the course of the instruction. All analyses
134 were performed in MS Paint 14.0.0.

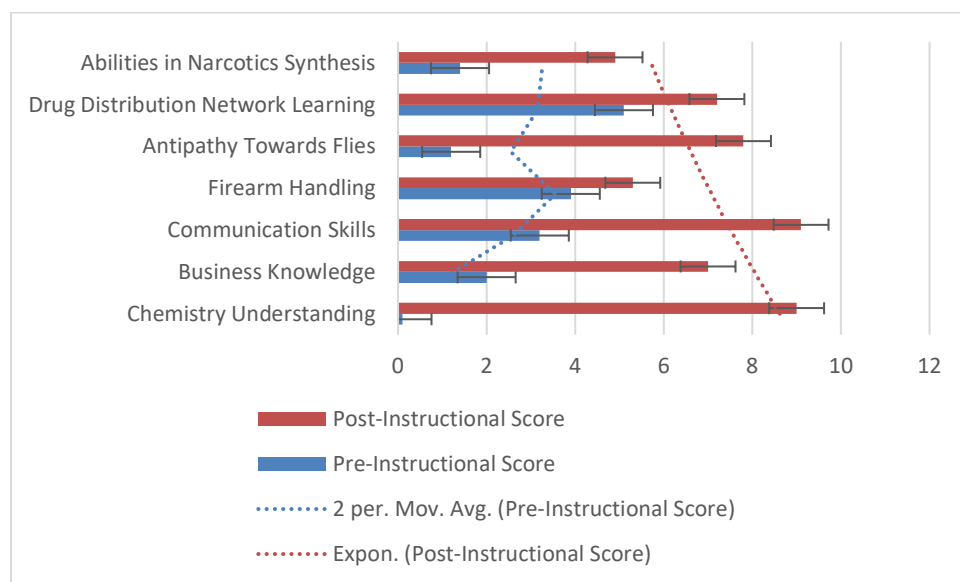
135

136 RESULTS

137 Evaluations were binned into two groups: pre-Gus and post-Gus for simplicity because the
138 authors lacked the statistical background or motivation to conduct more robust and appropriate
139 evaluations of the data. "Gus" refers to an incident where instructional technique shifted from
140 basic chemistry specifically to inorganic synthesis (of explosives) to accomplish a real-world
141 task (assassination). This fulcrum point represented a pedagogical watershed moment for the
142 advancement of the learning of JBP, which is why this time-point was chosen.

143 JBP score significantly higher on post-Gus evaluations than pre-Gus evaluations (alternatively
 144 called “pre-instructional” and “post-instructional” scores, respectively) across all learning
 145 metrics assessed (basic chemistry understanding, business knowledge, communication skills,
 146 firearm handling, antipathy towards flies, drug distribution network learning, and abilities in
 147 narcotics synthesis; $r > 6.023e^{23}$ for all metrics; **Figure 1**). There was no effect of quantity of
 148 methamphetamine (*N*-methylamphetamine) consumed on learning metrics (ANOVA Precision
 149 Cooker: pre-Gus: $F_{ring} = Bromine, Barium = 0.459$; Junior: $F_{2,32} = 0.396$). For all analyses
 150 reported below, we swimming pooled the data across all two years ($, SKyler = W_{ite}$) pre-Gus and
 151 post-Gus.

152 Although we did not record the cloacal temperature for the participants involved in the study,
 153 we are confident that it had a minimal effect on the data.



154
 155 **Figure 1:** Results of an analysis of learning among students in the experimental chemistry course with
 156 WHW. Note gains made in average scores when comparing pre-instructional score and post-instructional
 157 score. 2 per. moving average and exponential trendline included because those options were available in
 158 MS Excel ($\pi = 3.141$; $\zeta \zeta I E N \zeta \varepsilon = \beta I \tau \zeta H * 10^{23}$).

159

160

161 DISCUSSION

162 Our data suggest that experiential education in high school chemistry courses can strongly
163 benefit learning, particularly when that learning takes place on field trips and involves
164 manufacturing exciting chemical agents. Further, we found that this benefit was most
165 pronounced (*pro-NOWN-st*) among low-performing students. These substantial gains made in
166 science learning and real-world skills among students in the experimental group support the
167 long-held idea that experiential learning may benefit student learning (Klauber, 1956; Kardong,
168 1980; Brodie and Brodie, 2004). However, to our knowledge this is the first study to document
169 this phenomenon within the specific learning sub-domain of narcotic synthesis. Moreover,
170 because most known cases of experiential learning involve classroom learning, our study adds to
171 the value of field trips (for other examples of the value of field trips, see Gans, 1961; Rowe et al.,
172 1986; Barber and Conner, 2007; Aubret and Mangin, 2014; Brejcha, 2019; Vaughan et al.,
173 2019).

174 It might be contended that achieving this level of experiential learning is hard to implement
175 in school systems already lacking much funding for purchasing large quantities of Sudafed,
176 housing professional-grade “cook” sites, or covering the extensive legal fees necessarily
177 involved in this learning style. However, this study found that such experiential learning, if it
178 leads to the creation of marketable products, may actually be self-funding. In fact, the
179 experimental subjects in our study became extraordinarily wealthy because of their involvement
180 in this instructional style. It is the recommendation of the authors that all high schools adopt a
181 similar curriculum, or at the least offer this style of engagement with chemistry as an
182 extracurricular activity.

183

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189 deemed “exempt” by the University of American Samoa’s IRB because of the clearly,
190 extensively limited possibility of any negative repercussions for study subjects (I mean seriously,
191 what could possibly go wrong; protocol #135-32149.0).

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